TRITIUM HANDLING PRECAUTIONS

PHYSICAL DATA
Maximum Beta Energy: 0.019 MeV (100%)\(^{(1)}\)
Maximum Range of Beta in Air: 4.7 mm (0.19 in.)\(^{(2)}\)

OCCUPATIONAL LIMITS\(^{(3)}\)
Annual Limit on Intake: 80 mCi (3 GBq)
Derived Air Concentration: \(2 \times 10^{-6} \mu\text{Ci}/\text{mL} (740 \text{ kBq/m}^3)\).

DOSIMETRY
Millicurie (37 MBq) quantities of tritium do not present an external exposure hazard because the low-energy beta emitted cannot penetrate the outer dead layer of skin. Exposure to an atmosphere containing tritiated water results in intake of \(^3\text{H}\) by both inhalation and absorption through the intact skin\(^{(4)}\). Three to four hours after intake, ingested, inhaled or absorbed tritiated water is uniformly distributed in all body water\(^{(4)}\). On average, tritiated water is eliminated with a 10-day biological half-life\(^{(4)}\). Elimination rates can be increased by increasing water intake\(^{(5)}\).

DECAY TABLE
Tritium Physical Half-Life: 12.28 Years\(^{(6)}\)
To use the decay table, find the number of years in the left hand column and the number of months along the top of the chart, then find the corresponding decay factor. To obtain a precalibration number, divide by the decay factor. For a postcalibration number, multiply by the decay factor.

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</table>

\[^{(1)}\] Maximum Beta Energy: 0.019 MeV (100%)

\[^{(2)}\] Maximum Range of Beta in Air: 4.7 mm (0.19 in.)

\[^{(3)}\] Annual Limit on Intake: 80 mCi (3 GBq)

\[^{(4)}\] Derived Air Concentration: \(2 \times 10^{-6} \mu\text{Ci}/\text{mL} (740 \text{ kBq/m}^3)\)

\[^{(5)}\] On average, tritiated water is eliminated with a 10-day biological half-life. Elimination rates can be increased by increasing water intake.

\[^{(6)}\] Tritium Physical Half-Life: 12.28 Years
**GENERAL HANDLING**

**PRECAUTIONS FOR TRITIUM [³H]**

1. Designate areas for handling ³H and clearly label all containers.
2. Prohibit eating, drinking, smoking and mouth pipetting in room where ³H is handled.
3. Use transfer pipettes, spill trays and absorbent coverings to confine contamination.
4. Handle potentially volatile compounds in ventilated enclosures.
5. If enhanced containment is necessary, handle volatile compounds in closed systems vented through suitable traps.
6. Sample exhausted effluent and room air by continuously drawing a known volume through a membrane filter followed by an impinger containing water.
7. Wear disposable lab coat, gloves and wrist guards for secondary protection.
8. Select gloves appropriate for chemicals handled.
9. Maintain control by regular monitoring and prompt decontamination of gloves and surfaces.
10. Use open-window ionization detector or liquid scintillation counter to detect ³H.
11. Submit periodic urine samples for bioassay to determine uptake by personnel.
12. Isolate waste in sealed clearly labeled containers according to approved guidelines.
13. Establish air concentration, surface contamination, and bioassay action levels below regulatory limits. Investigate and correct any conditions that may cause these levels to be exceeded.
14. On completing an operation, secure all ³H; remove and dispose of protective clothing and coverings; monitor and decontaminate self and surfaces; wash hands and monitor them again.

Many tritium compounds readily penetrate gloves and skin. Handle these compounds remotely, wear two pairs of gloves and change the outer layer at least every 20 minutes. Tritiated DNA precursors are considered more toxic than tritiated water depending on their route of intake\(^{(4)}\). However, they are generally less volatile and do not normally present a significantly greater hazard.

**REFERENCES**