

# ***IN SITU* COMMUNITY CONTROL OF THE STABILITY OF BIOREDUCED URANIUM**

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**NABIR**

**UT CBA**

We could not do this work without  
our excellent collaborators!

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and Derek R. Lovley

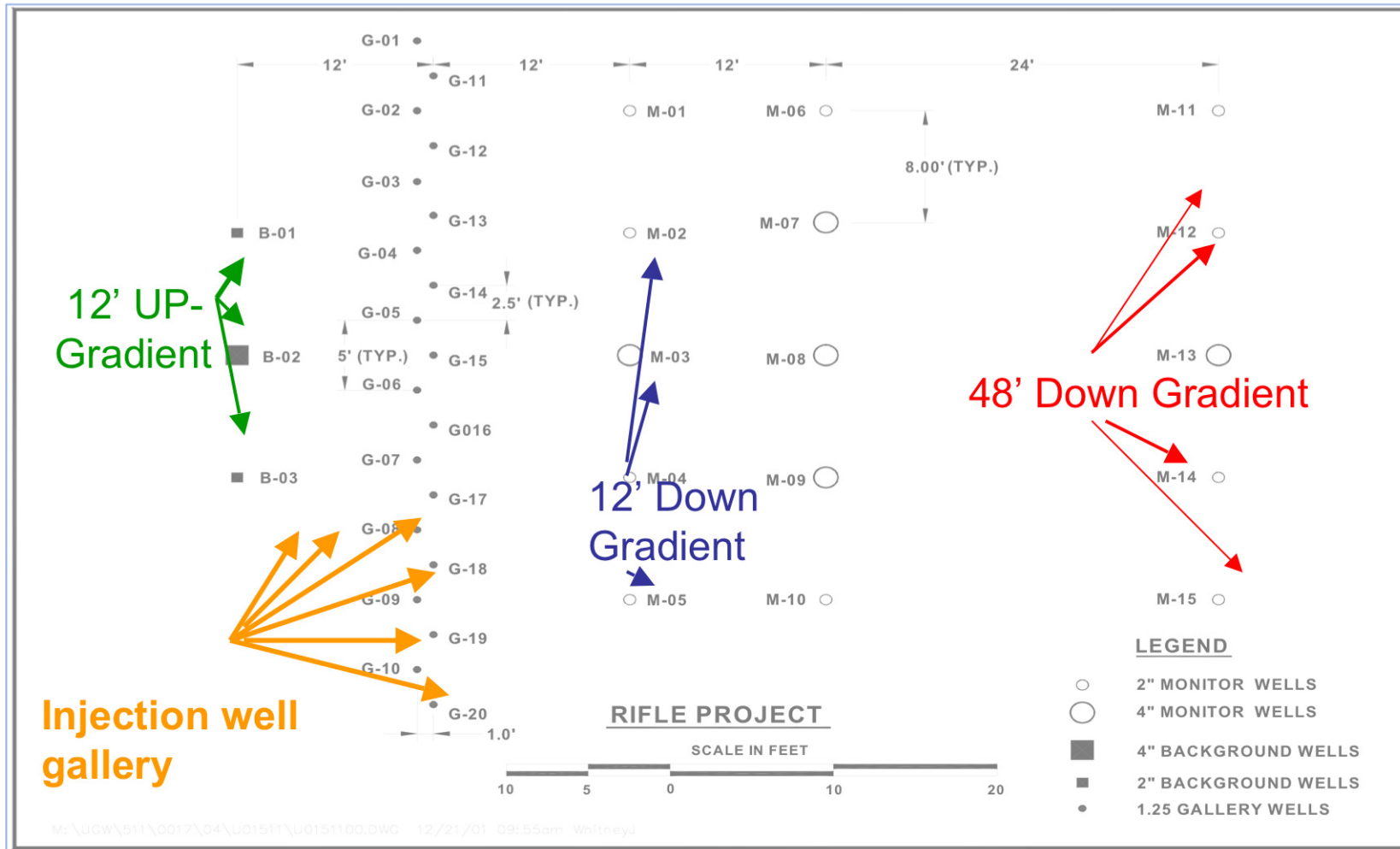
**UMASS**

Richard Dayvault  
**S.M. Stoller Corp.**

Darrell Chandler  
**ANL**

# Shift of viable biomass & community composition by acetate addition to groundwater ~UMTRA site

## Well field



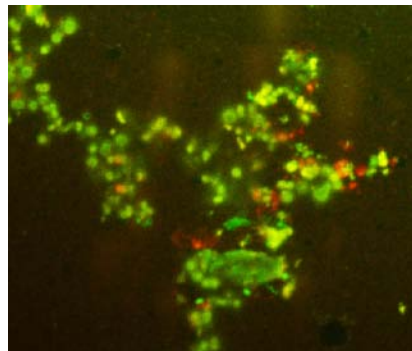
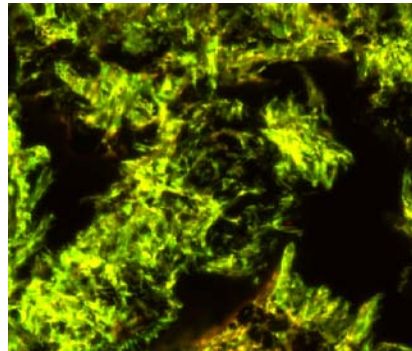
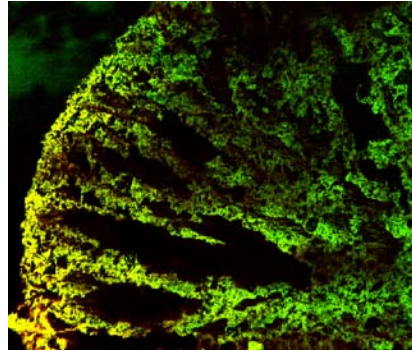
# UMTRA Old Rifle site



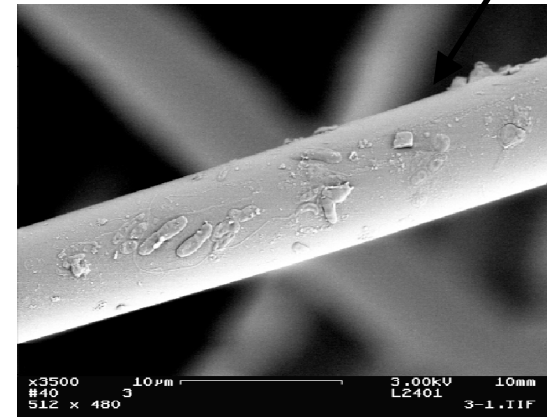
Sampler with Beads

CLSM image of microbial colonization live/dead stain

## Confocal Laser Scanning Microscopy of beads



SEM of microbial colonization of glass wool



## Acetate Infusion induced

A. Changes in the mineralogy   $\text{FeS}_{0.9}$

B. Changes in the microbial community

Program is to test hypotheses as to **How changes in A & B Maintained the decrease in U(VI) in the groundwater**

### **TEST**

Monitor U(VI) loss in-well sediment incubators  
different mineral & microbiological amendments

# Assessing Subsurface Microbiota

## Collecting Microbes

### Bead Coupons

Must colonize from water

Actively Growing

$^{13}\text{C}$  in DNA & PLFA

Biomass Intermediate

Complexity Increases

### Groundwater

Grab Sample

mRNA

+ Non Growing

Lowest

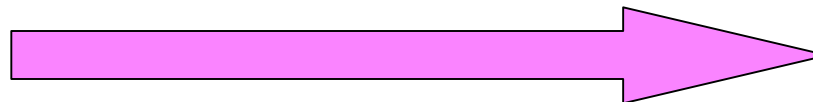
### Sediment

Disruptive

Includes slow & non

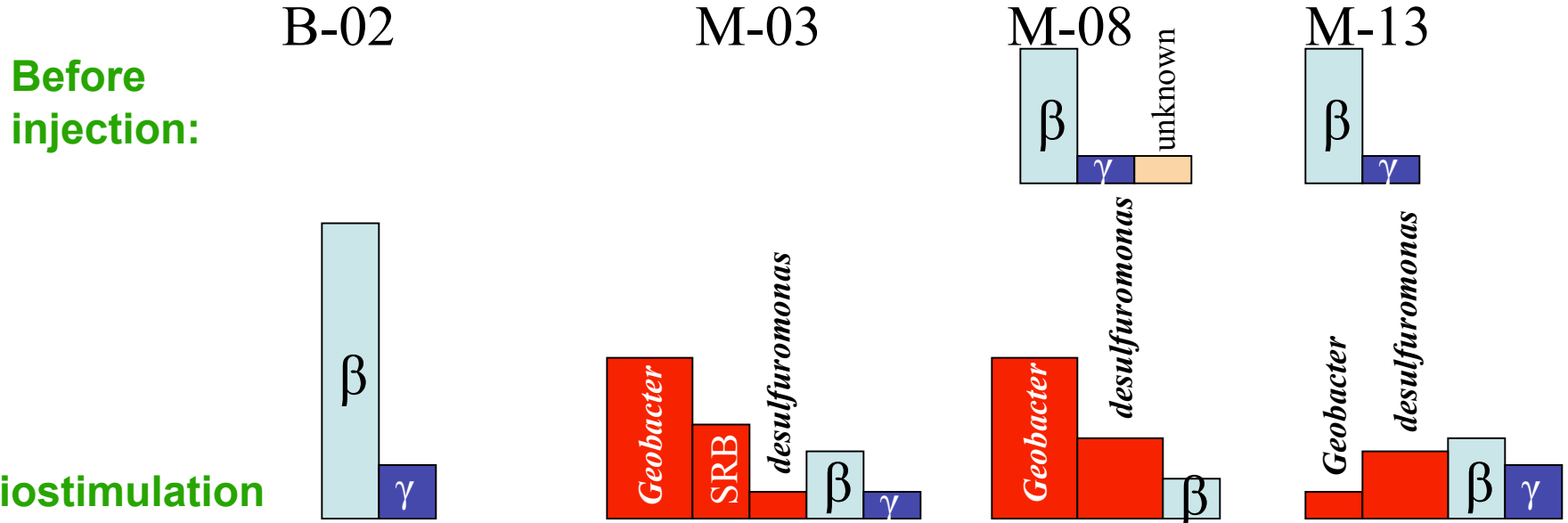
Growing

Highest



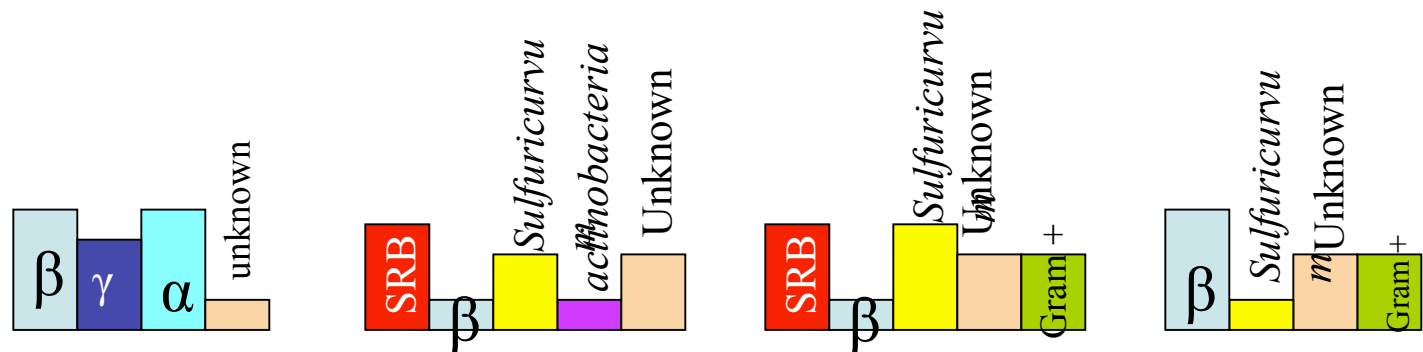
# Why did U(VI) decrease in groundwater persist 7-9 months after acetate infection stopped? (Bead Coupons)

Up-Gradient 12' Down-Gradient 24' Down-Gradient 48' Down-Gradient



\*Red--- denotes  $\delta$ -proteobacteria

**7~9 months after acetate addition stopped**



## Preliminary data (DGGE from Bead Coupons) suggests:

**Up gradient & before injection stable community** dominated by  $\beta$ -proteobacteria (*Hydrogenophaga*, *Dechloromonas*, *Rhodoferax*, *Ferribacterium*, *Rhodocyclus*, *Methylophilus*, *Azoarcus*,  $\gamma$ -Proteobacteria *Pseudomonas*

**A Gram-negative, heterotrophic, carbon-limited, (high cyclo/monoenoic PLFA), facultative-anaerobic, oxygenated (UQ/MK~2), utilizing refractory organics**

**Infusion** *Geobacter*, *Desulfuromonas/Pelobacter* DIRB,  $\delta$ -Proteobacteria *Desulfobacter*,  $\beta$ -proteobacteria *Ferribacterium*,  $\gamma$ -Proteobacteria *Pseudomonas*,

**Rapid Biostimulation** DIRB,SRB goes anaerobic & growth stimulated

**7-9 months later** , *Sulfuricurvum*,  $\beta$ -&  $\gamma$ -Proteobacteria, Gram-positive Clostridia,  $\delta$ -Proteobacteria SRB

**Sulfur oxidizing bacteria** use sulfide to ➡ reduce Nitrate/ Oxygen maintain anaerobic status UQ/MK ~06-0.2 & SRB form U(IV)?



## Hypotheses:

- 1) After stimulation Fe(III) terminal electron acceptor ↓  
non sulfate-reducing DIRB will be leave or be out-competed  
by more versatile microbes ? SRB
- 2) SRB & Sulfur oxidizing bacteria play a critical role in the post-treatment maintenance of bio-reduced uranium by directly  
reducing U(VI), generating  $H_2S$ ,  $HS^-$  and/or  $FeS_{0.9}$  ➔ oxygen  
sinks maintaining U(IV).
- 3) Bioprecipitated amorphous  $FeS_{0.9}$  in sediments will maintain  
low U(IV) reoxidation rates under conditions of low biomass,  
especially in presence **of sulfur oxidizing bacteria**

but  $FeS_{0.9}$  by itself is not sufficient to remove U(VI) from  
groundwater by abiotic reduction

## Conceptual Model

**Start Acetate** to C-starved  $\beta$ -proteobacter, oxygenated  $\Rightarrow$   
Anaerobic  $\Rightarrow$  Geobacter + DIRB  $\Rightarrow$  reduce Fe(III)  $\Rightarrow$  great  $\uparrow$   
increase biomass  $\Rightarrow$  Lo  $\downarrow$  cal Fe(III) surfaces all reduced  $\Rightarrow$  wave  
Geobacter (+ DIRB) moves distally

Continued acetate  $\Rightarrow$  SRB increase in diversity (DSR) & biomass  
 $\Rightarrow$  produce  $\text{HS}^-$   $\Rightarrow$   $\text{HS}^- + \text{Fe(II)} \Rightarrow \text{FeS}_{0.9}$

**Stop Acetate** Biomass  $\downarrow$  Cell lysis feeds Heterotrophs

Gram-positive Clostridia + Desulfotomaculum (+ SRB)  $\Rightarrow$  reduce  
U(VI) & produce acid  $\downarrow$  carbonate & U(VI) complex

Sulfur Oxidizing (Sulfuricurvum) use  $\text{HS}^-$  to reduce  $\text{NO}_3$  &  $\text{O}_2$

Without  $\text{NO}_3$  &  $\text{O}_2$  Dechloromonas & Geobacter not reoxidize  
U(IV)

Does Cigar Lake U mine deposit have high  $\text{FeS}_{0.9}$  + SOB & low  
UQ/MK?

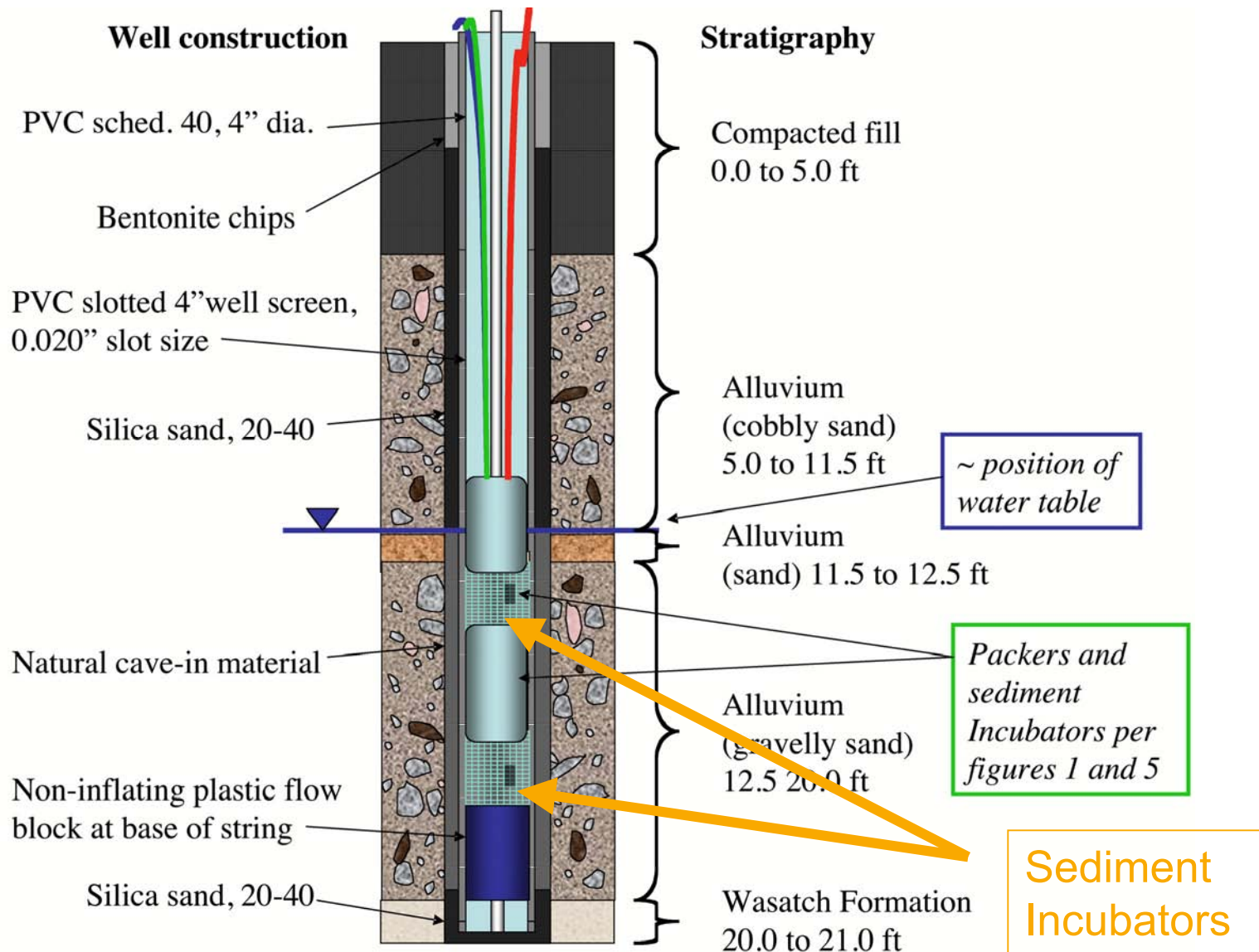


Figure 4. Proposed emplacement of two sediment incubators in a typical 4" well (B-02, Old Rifle UMTRA Site).

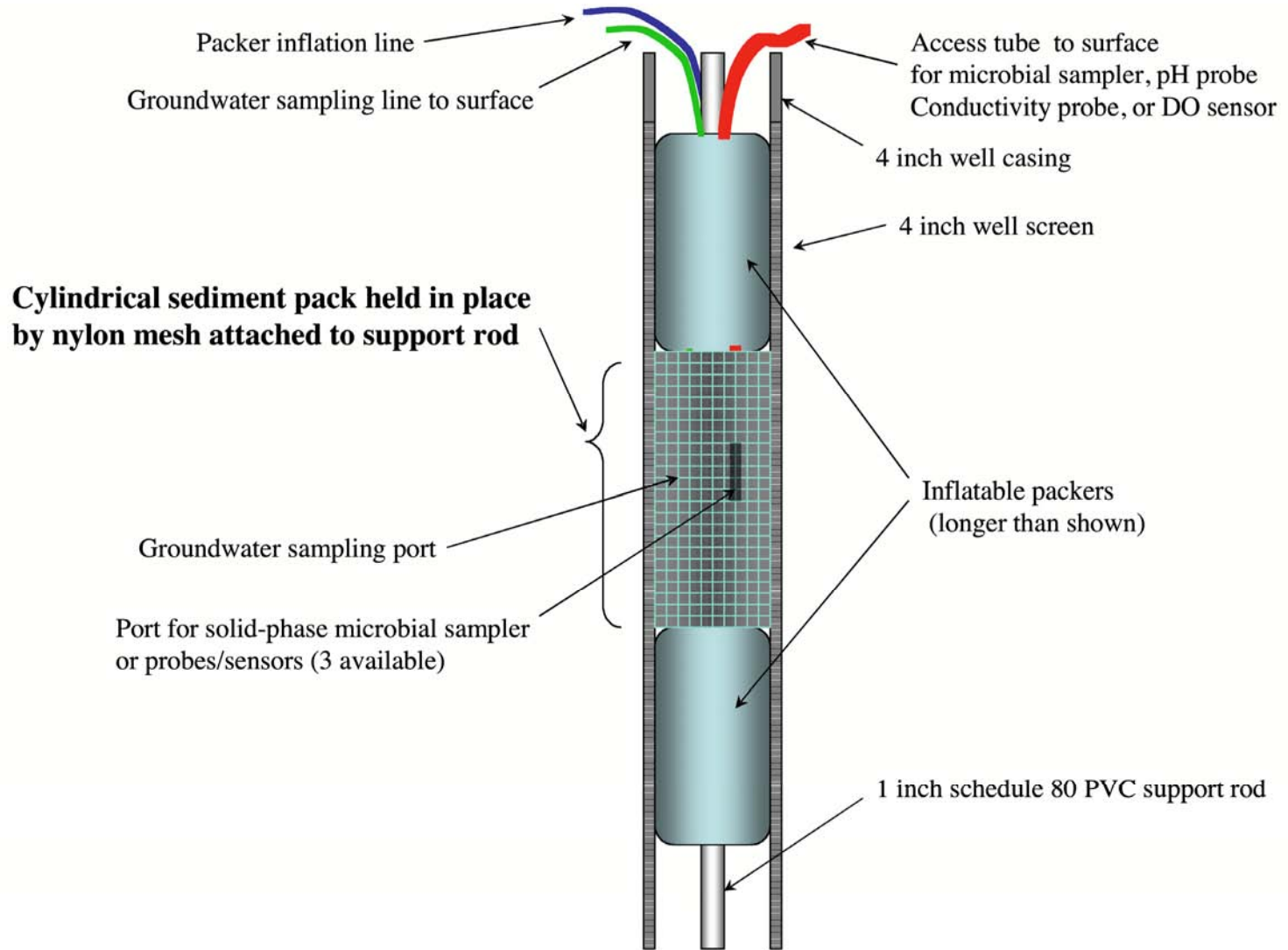


Figure 1. Conceptual design of sediment incubator

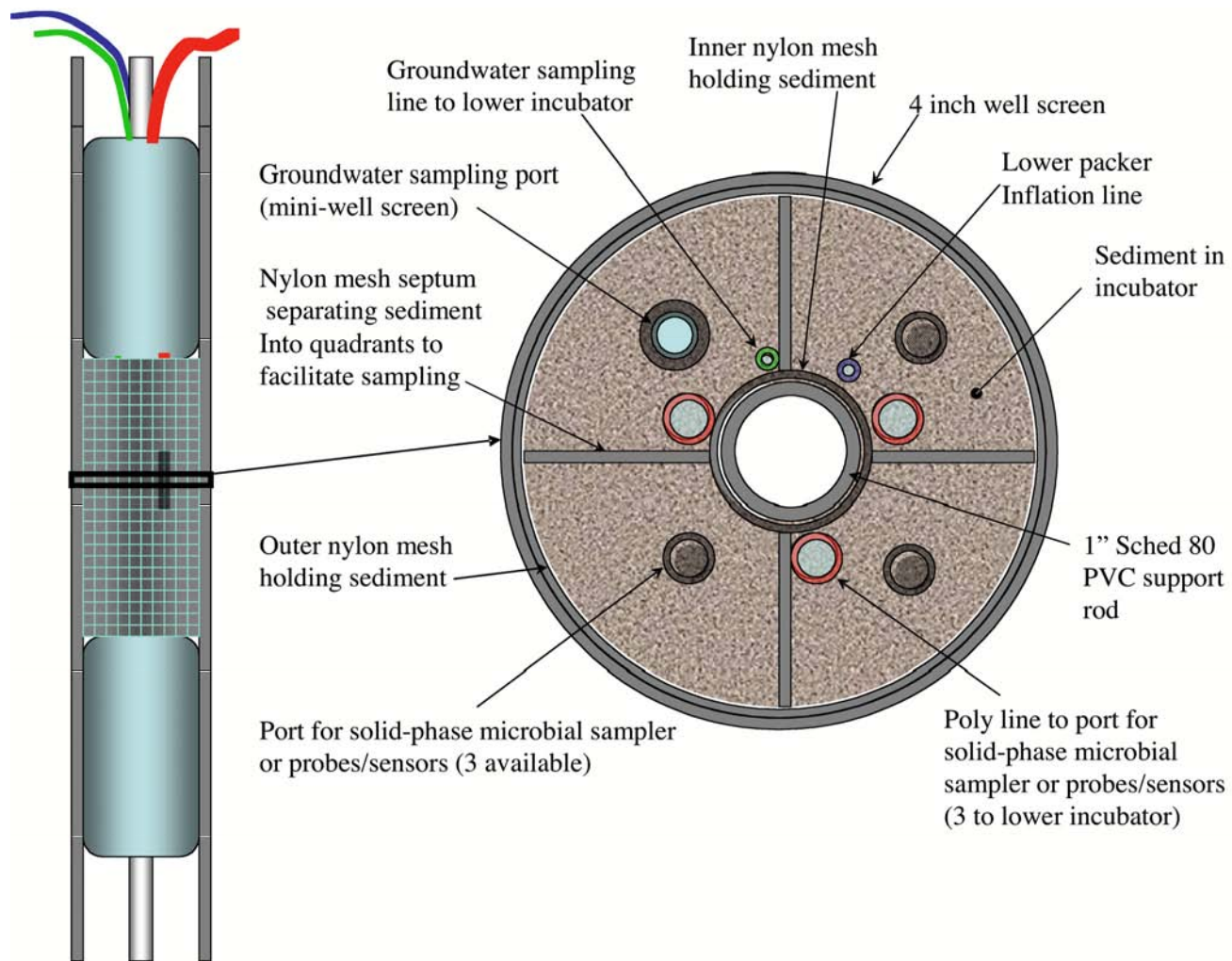


Figure 5. Top view (section) of sediment incubator.

Sediment for the in situ sediment incubators

Rifle sediment      Cobs removed & mixed 1-3 mm sieved for higher permeability sand-silt

U(VI) 0.17 mg/L from groundwater during perfusion

Sediment + DIRB + Lactate ➡ SRB↑ DIRB ↓ [Hypothesis 1]

Sediment + Lactate + DIRB vs Sediment + Lactate + SRB  
measure bioreduced U(IV) [Hypothesis 2]

Sediment + FeS<sub>0.9</sub> Sterile (short time) vs SRB + acetate  
[Hypothesis 3]

**Model: Sediment + FeS<sub>0.9</sub> + Lactate SRB + SOB**

# Assessing subsurface microbiota

## From Sediment Samplers

DNA 16S rRNA, rDNA,

Genes

DSR (SRB),

DIRB ? NADPH-iron reductase?

soxA sulfite oxidase

by DGGE, Q-PCR, T-RFLP

RNA D. Chandler

Lipids PLFA, Respiratory Quinones, PHA, DMA (Clostridia) ? Spores (DPA)

## Better Respiratory Quinone Assessment

Problem present at mmol/mol PLFA

### HPLC/electrochemical cell/electrospray/ MS/MS

HPLC separates components so greater duty cycle

Electrochemical cell ➔ Reduces only Quinones at the specific  $E_0$  potential : + 112 mV Ubiquinone; + 36 mV Desmethyl Menaquinone;  
- 74 mV Menaquinone

100% ionized with - 2 charge ➔ ideal for electrospray ionization

Compare to atmospheric pressure chemical ionization (APCI)  
~inefficient  $H^+$  charge transfer from activated gas

MS/MS search for progenitors ions at products: UQ m/z – 197, - 98.5  
DMK m/z - 173, -86.5; MK m/z -187 -93.5

**Greatly increase sensitivity ➔ greatly increase localization**