### PNNL/Alabama/ORNL Project Activities and Results

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Pacific Northwest National Laboratory Operated by Bathelle for the U.S. Department of Energy

# **Other Contributors**

NABIR FRC Staff (well construction and sediment collection, groundwater analyses)

- ORNL:
  - Wiwat Kamolpornwijit (groundwater sampling and analysis)
  - Melanie Mayes (intact core excavation and operation)
  - Young-Jin Kim (uranium sorption and transport studies)
- University of Alabama
  - Ken Overstreet (sediment collection and core logging)
  - Santosh Mohanty (laboratory slurry experiments)
- ► PNNL:
  - Yilin Fang (reactive transport modeling)
  - Frank Spane (hydraulic test design and analysis)



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# **Fractured Saprolite**



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# Problem

▶ 10-20% of porosity is accessible by active pumping

- 10-20% of porosity will respond to pumping on intermediate time scales
- 60-80% of porosity (and associated contamination) is hydraulically inaccessible (controlled by slow diffusive mass transfer)

# Hypothesis

Mobile radionuclides in low-permeability porous matrix regions of fractured saprolite can be effectively isolated and immobilized by stimulating localized in-situ biological activity in highly-permeable fractured and microfractured zones within the saprolite.



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### **Progress Areas**

Field Site Development and Characterization
 Laboratory Fe/U Reduction Potential Analyses
 Bench-Scale Proof-of-Principle
 Numerical Model Application

# **Field Site Development**





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Bedrock (Nolichucky Shale / Limestone)

# Exploratory Wells FW201, FW202May 2003



### Well Installation – Round 1





# Round 2 Well Installation Detailed Characterization Wells August 2003



### **Groundwater Constituents**

- U in µM range
  pH circumneutral
  Nitrates as high as ~125 mM !
  - Sedimentassociated U low (<< 1 µmol/g)</li>

Date	Well	pH	NO3- (ppm)	SO4= (ppm)	U (ppb)
08/26/03	FW204	7.30	237.4	118.6	547.4 441.1
	FW205	6.87	495.0	65.8	77.4 74.1
	FW206	6.59	684.2	34.4	15.7 15.6
	FW207	6.83	368.2	48.9	180.5 150.6
	FW208	5.48	3077.7	11.3	69.3 67.4
	FW209	6.26	1078.2	13.4	239.7 238.8
	FW210	6.36	7719.5	12.0	185.7 117.6
	FW211	6.33	1584.3	62.1	421.8 428.8

Note: Top number for U is unfiltered; bottom is filtered.

### Sediment Extractions FWB201 and FWB202



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### Hydraulic Testing – Deep Zone

- Installed in-well pressure transducers with dataloggers in several wells
- Tracked ambient water level fluctuations in response to rainfall, barometric pressure, seasonal variations
- Tracked water level response to pumping in FW207.





# Hydraulic Testing – Deep Zone

### Conclusions:

- Barometric response in lower zone strong, can be accounted for. Nature of barometric response is consistent with double-porosity conceptual model.
- Estimated  $K = 4x10^{-7} \text{ m/s}$
- Primary connectedness in E-W direction (parallel to geologic strike)
- Low sustainable pumping rates (tens of ml/min)
- Estimated direction of maximum head gradient ~ 30 degrees west of south.

# **Disturbed or Undisturbed Saprolite?**

### Factors:

- Lower levels of uranium contamination
- Much higher levels of nitrate
- Low water movement rates
- Conclusion: Shift focus to upper zone (disturbed saprolite or "fill" zone).
  - Structure less well-defined, but still prevalent
  - Higher levels of U(VI)
  - Low levels of nitrate
  - Significant sustainable pumping rates

### Round 3 Well Installation – Feb. 2004

#### Purposes:

- Confirm head gradient direction
- Prepare for tracer injection to establish flow direction
- Construct wells to be used as part of biostimulation flow cell
- Refine understanding of contaminant distribution



~10 m

### **Aqueous Chemistry**

30020-

► U(VI)<sub>aq</sub> ~ 4-5 µM

![](_page_19_Picture_2.jpeg)

OP05 1.5 30000 29980 OTPB29 @FW217 29960 OP06 895,9% FW218 29940 FW201-2 GW835 29920 51260 51280 51300 51320 51340 51360 51380 51400 51420 30020-DP05 30000 29980 TPB29 ●FW217 29960 PR0200 29940 W201-2 7月月月1 03W835 29920-51280 51300 51320 51360 51400 51420 51260 51340 51380

▶ NO3 ~ 0.5 mM

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OP04

### **Azimuthal Resistivity Survey**

#### Deep zone

constant of the

#### Shallow zone

![](_page_20_Figure_3.jpeg)

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### **Next Steps**

- Hydraulic characterization of undisturbed zone
  - Slug interference testing
  - Tracer test
- Completion of flow cell infrastructure
  - Additional injection/monitoring wells
  - Geophysical tomography wells on perimeter

![](_page_21_Figure_7.jpeg)

### Laboratory Fe/U Reduction Potential Experiments

Sediment laboratory wetchemical analysis and molecular analysis of sediment and groundwater samples

![](_page_22_Picture_2.jpeg)

### **Results from Intact Saprolite**

Area 2 Sediment (23-26 ft) slurry incubations (twelve samples each with and without 10 mM ethanol)

![](_page_23_Figure_2.jpeg)

Variability associated with non-uniform solid:solution ratio in samples

Normalizing to total Fe shows trend of increased reduction in +Ethanol flasks

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![](_page_24_Figure_0.jpeg)

![](_page_24_Figure_1.jpeg)

\* Uranyl acetate added at day 96; U(VI) and U<sub>Resid</sub> measured at day 161

# **Next Steps**

Sediment from disturbed saprolite zone (most recent well installation) has been obtained

New slurry experiments will be initiated shortly.

### Element 3: Bench-scale testing of hypothesized process

Undisturbed column studies

- Perform proof-ofprinciple experiment at bench scale.
- •Derive rate constants for use in upscaling studies.
- Identify reaction network and examine solid phase reaction products.

![](_page_26_Picture_5.jpeg)

![](_page_27_Picture_0.jpeg)

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### **Intact Column Studies**

![](_page_28_Picture_1.jpeg)

![](_page_28_Picture_2.jpeg)

![](_page_29_Picture_0.jpeg)

Amorphous Fe (ammonium oxalate extraction)  $11.0 \pm 0.2$  mmole/kg Total Fe (citrate-dithionate-bicarbonate)  $320 \pm 10$  mmole/kg Total Mn (acidic hydroxylamine hydrochloride)  $3.1 \pm 0.1$  mmole/kg

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# **Status and Next Steps**

- Three cores have been collected and set up in the laboratory
- The columns have been fully water saturated and are being partially desaturated under controlled vacuum
- Loading of micropores with uranium under vacuum will be started shortly
- Three different treatments will be applied
  - Control (tracer only)
  - Tracer plus electron donor
  - Tracer, electron donor, and electron shuttle

### Batch U(VI) Sorption Studies FRC Background Sediments

![](_page_31_Figure_1.jpeg)

 $C_0 = 1 \ \mu M$ 

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### Batch U(VI) Sorption Studies FRC Background Sediments

![](_page_32_Figure_1.jpeg)

 $C_0 = 10 \ \mu M$ 

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### Effect of Solid:Solution Ratio FRC Background Sediments

![](_page_33_Figure_1.jpeg)

#### Battelle

### Site Affinity Distributions Background Area Sediments

![](_page_34_Figure_1.jpeg)

### U(VI) Sorption Edges Background Area Sediments

![](_page_35_Figure_1.jpeg)

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### U(VI) Transport through Repacked Column Background Area Sediments

![](_page_36_Figure_1.jpeg)

![](_page_36_Picture_2.jpeg)

### Numerical model application

• Related NABIR research (Roden/Burgos) has developed reactionbased models of coupled iron(iii) oxide and U(VI) reduction kinetics including sorption of Fe(II) and U(VI) and site blocking.

![](_page_37_Figure_2.jpeg)

### **Numerical Model Application**

 We have developed a 2D simulation framework with heterogeneous aquifer properties.

![](_page_38_Figure_2.jpeg)

![](_page_38_Figure_3.jpeg)

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# U(VI) Sorption – 1000 days

![](_page_39_Figure_1.jpeg)

Aqueous U(VI)

![](_page_39_Figure_2.jpeg)

![](_page_39_Figure_3.jpeg)

**Battelle** 

Acetate

![](_page_40_Figure_2.jpeg)

![](_page_40_Figure_3.jpeg)

FeOOH<sub>s</sub>

**Battelle** 

Acetate

Fe(II)<sub>aq</sub>

![](_page_41_Figure_2.jpeg)

![](_page_41_Figure_3.jpeg)

**Battelle** 

Biomass<sub>s</sub>

**Biomass**<sub>aq</sub>

![](_page_42_Figure_2.jpeg)

![](_page_42_Figure_3.jpeg)

**Battelle** 

![](_page_43_Figure_1.jpeg)

**Battelle** 

# **Next Steps**

Link fracture flow and reactive transport models

- Apply model to FRC site (Area 2)
- Incorporate site-specific heterogeneity information as it becomes available
- Use for field-scale experimental design

### For More Information...

 Project website (contact <u>tim.scheibe@pnl.gov</u> for access information)

NABIR FRC website (http://www.esd.ornl.gov/nabirfrc/)

![](_page_45_Picture_3.jpeg)

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